

EVT 2011_12_02AM

Bacterial Protein Fingerprinting

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Bacterial Protein Fingerprinting



Experiment

There is enough of each sample for six (6) groups sharing three polyacrylamide gels.

Upon receipt, refrigerate component E and place F - H in the freezer.

All components are intended for educational research only. They are not to be used for diagnostic or drug purposes, nor administered to or consumed by humans or animals.

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EXPERIMENT COMPONENTS

- A Escherichia coli slant
- B Serratia marcescens slant
- C Micrococcus luteus slant
- D Bacillus subtilus slant
- E Tris-EDTA-Glucose (TEG) buffer
- F Lysozyme
- G Protein sample buffer
- H Unknown (ready for electrophoresis)
- Tris-glycine-SDS electrophoresis buffer (10x)
- Protein InstaStain®
- Practice gel loading solution
- ReadyPour™ agar
- Nutrient Broth
- Screw cap tubes
- Sterile inoculating loops
- Sterile 1 ml pipets
- Sterile petri plates, 100 x 15 mm

None of the components have been prepared from human sources.

REQUIREMENTS

- MV10 or MV20 Vertical electrophoresis apparatus
- D.C. Power Supply
- Three 12% precast SDS polyacrylamide gels (Cat. #651 or #652)
- Micropipet and tips
- (Cat. #638, Fine Tip Micropipet Tips recommended)
- Microcentrifuge
- Incubation oven (37° C)
- Microcentrifuge tubes
- Hot plate, Microwave oven or Bunsen burner
- White light box
- Glass staining trays
- Distilled water
- Glacial acetic acid
- 95 100% Methanol



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<u>Storage</u>

Room temperature Room temperature Room temperature Room temperature Refrigerator Freezer Freezer Freezer

Room temperature Room temperature Room temperature Room temperature Room temperature



Background Information

BACTERIAL PROTEIN FINGERPRINTING

Bacteria have been historically identified and classified according to morphological traits such as shape, size, motility, gram stain, and macroscopic growth characteristics. However, similar morphological characteristics are shared by many bacteria. Biochemical, nutritional and physiological traits are very important in bacterial identification and classification. These traits include oxygen requirements, glucose fermentation, pH response, ability to grow on selective media, ability to cause hemolysis, and response to antibiotics. Bacterial evolutionary descent and phylogenetic classification are tractable by the study of bacterial DNA, RNA, and proteins. The approach to bacterial relationships by using molecular phylogenetic data is relatively new. The use of molecular signatures such as ribosomal RNA is useful in the identification of different bacterial species. Bacterial cells, such as E.coli, contains approximately 2000 different kinds of proteins. The amount and types of proteins within the cell can vary depending on the environmental and physiological conditions. While many proteins are physico-chemically and functionally similar between different bacteria it is unlikely that all of them will share the same distribution of molecular weights, shapes, charge and immunogenicity. Furthermore, the relative concentrations of proteins that are otherwise similar in other characteristics can vary between two different types of cells and can be used as a fingerprint pattern.



Figure I - The chemical structure of sodium dodecylsulfate (SDS).

POLYACRYLAMIDE GEL ELECTROPHORESIS

Sodium dodecylsulfate (SDS) polyacrylamide gel electrophoresis is particularly useful to analyzing the complex profile created by a total soluble protein lysate. Total bacterial protein lysates analysis by SDS electrophoresis is frequently done to check for the over expression of protein. Sodium dodecylsulfate (SDS) is a detergent which consists of a hydrocarbon chain bonded to a highly negatively charged sulfate group (Figure 1).

SDS binds strongly to most proteins and causes them to unfold to a random, rod-like chain. No covalent bonds are broken in this process. Therefore, the amino acid composition

and sequence remains the same. Since its specific three-dimensional shape is abolished, the protein no longer possesses biological activity. Proteins that have lost their specific folding patterns and biological activity but have their polypeptide chains remaining intact are called denatured. Proteins which contain several polypeptide chains that are associated only by non-covalent forces will be dissociated by SDS into separate, denatured polypeptide chains. Proteins may contain covalent cross-links known as disulfide bonds. These bonds are formed between two cysteine amino acid residues that can be located in the same or different polypeptide chains. High concentrations of reducing agents, such as 2-mercaptoethanol, wll break disulfide bonds. This allows SDS to completely dissociate and denature the protein. Proteins that retain their disulfide links bind less SDS, causing anomalous electrophoretic migration.

In most cases, SDS binds to proteins in a constant ratio of 1.4 grams of SDS per gram of protein. On average, the bound SDS molecules is half the number of amino acid residues



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Background Information

in the polypeptide. The large quantity of bound SDS efficiently masks the intrinsic charges in the protein. Consequently SDS denatured proteins are net negative and the binding of the detergent is proportional to the mass of the protein. The charge to mass ratio in denatured proteins is constant and the shapes of SDS denatured proteins are all rodlike. The larger the molecular weight of the protein the longer the rod-like chain. During SDS electrophoresis, proteins migrate through the gel towards the positive electrode at a rate that is inversely proportional to their molecular weight. The pores in the gel distinguish these size differences, the smaller the protein, the faster it migrates. The molecular weights of unknown proteins are obtained by comparison of the relative positions after electrophoresis to the positions of standard SDS denatured proteins electrophoresed in parallel.

The electrophoretic mobility of the proteins is also affected by the gel concentration. Higher percentage gels are more suitable for the separation of smaller proteins and of peptides. The pore size in polyacrylamide gels is controlled by the gel concentration and the degree of polymer cross linking. The polyacrylamide gel is formed by mixing the monomer, acrylamide, the cross-linking agent, methylenebis-acrylamide, and a free radical generator ammonium persulfate, in aqueous buffer (Figure 2). Free radical polymerization of the acrylamide occurs. At various points the acrylamide polymers are bridged to each other.

It should be noted that acrylamide is a neurotoxin and can be absorbed through the skin. However, in the polymerized polyacrylamide form it is non-toxic. The polymerization process is inhibited by oxygen. Consequently, polyacrylamide gels are usually prepared between two glass plates separated by strips called spacers. As the liquid acrylamide mixture is poured between the plates, air is displaced and polymerization proceeds.

SURVEY OF BACTERIAL SAMPLES

Escherichia coli and Serratia marcescens are gram negative rods of the Enterobacteriaceae family. Members of this family generally can ferment glucose and other sugars, and require minimal media containing salts and small amounts of glucose. These bacteria are facultative anaerobes and do not form spores. Several members such as E.coli are found in the intestinal tracts of animals. *S.marcescens* is found in the soil or water. S.marcescens produces the red, pyrrole pigment prodigiosin at 25° C in starchy media.

Micrococcus luteus is gram positive and is a member of the Micrococcaceae family. The Staphylococcus genus is a well known, pathogenic member of this family. The family is characterized by spherically shaped cells that divide to form grape-like clusters. They do not form spores. Members of the genus *Micrococcus* are aerobes and are able to live under a wide variety of conditions in soil, dust, seawater and dairy products. *M.luteus* forms yellow pigments.

Bacillus subtilus is a gram positive, rod shaped member of the family Bacillaceae. Members of this family are endospore formers. B.subtilus is found in dust and hay. It is a strict aerobe. This bacteria produces the antibiotic subtilin and a closely related strain produces the antibiotic bacitracin. These antibiotics are peptides that interfere with the cell wall synthesis of mostly gram positive bacteria.





Background Information

The cell walls of gram negative bacteria consist of an outer lipid membrane containing glycolipids and lipopolysaccharides that project into the external environment. Beneath the outer membrane is the rigid meshwork of peptidoglycan. Beneath this peptidogly-can layer is the periplasmic space which contains proteins. The other side of the space is bounded by the cytoplasmic membrane. The peptidoglycan comprises 5-15% of the cell wall components by weight. The cell walls of gram positive bacteria tend to be much thicker. They do not possess an outer membrane and are structurally simpler. The peptidoglycan can be 20-80% of the cell wall components. The many layers of peptidoglycan in the gram positive bacteria form a relatively homogenous network.

Due to the outer membrane and its molecular projections the gram negative bacteria are much less susceptible to the activity of lysozyme than the gram positive bacteria. Lysozyme hydrolyzes the glycosidic bonds between the N-acetylglucosamine and N-acetylmuramic acid residues in the peptidoglycan. The viscosity increase is due to the release of high molecular weight DNA and other cell biomolecules. These effects are not readily observed after lysozyme treatment of gram negatives. Conversely, the addition of a detergent such as SDS to a suspension of gram negative cells results in cell lysis as evidenced by increased viscosity. Exposure of gram positive cells with intact peptidoglycan to SDS does not result in significant amounts of lysis as judged by viscosity.





Experiment Overview

EXPERIMENT OBJECTIVE:

The objective of this experiment is for students to prepare soluble protein lysates from several species of bacteria and analyze the electrophoretic protein profiles by identifying the fingerprint pattern of an unknown sample.

LABORATORY SAFETY

Gloves and goggles should be worn routinely as good laboratory practice.



Un-polymerized acrylamide is a neurotoxin and should be handled with extreme caution in a fume hood. Gloves and goggles must be worn at all times. Use a pipet pump to measure polyacrylamide gel components. Polymerized acrylamide precast gels are safe but should still be handled with gloves.







In this experiment, different species of bacteria will be grown on nutrient agar plates. The cells will be harvested and disrupted under denaturing conditions to obtain a total, crude protein lysate. SDS polyacrylamide gel electrophoresis is capable of resolving hundreds of denaturing protein components on the basis of their size. The bacteria used in this laboratory have characteristic protein electrophoretic profiles. Protein patterns of an unknown protein lysate will be compared in parallel with known samples prepared by the procedures outlined below. After electrophoresis the proteins will be visualized with stain and you will determine whether or not the unknown corresponds to one of the samples you have prepared.

GROWTH OF BACTERIAL CULTURES

First Day

1. Obtain two nutrient agar plates with different letter designations written on the bottom. Your instructor will determine which plates your group will get. Put your lab group number next to the letter on each plate. Each plate will be inoculated with a liquid suspension of the corresponding cells from one of the four bacterial slants identified in Step 4.

CHANGE PIPET TIPS BETWEEN EACH PLATE INOCULATION.

- 2. With a fresh, sterile pipet withdraw 0.1 ml of cell suspension from the lettered vial that corresponds to one of your plates. Deposit the liquid to the surface of the agar at the center of the corresponding plate.
- 3. Using a sterile loop thoroughly spread the liquid evenly over the entire surface of the agar plate. Rotate the plate and streak back and forth to obtain complete coverage. Do not apply too much pressure otherwise the agar may be damaged. Try to keep the plate partially covered while spreading or work under a flame. Cover the plate after spreading.
- 4. With a fresh, sterile pipet withdraw 0.1 ml of cell suspension from the lettered vial that corresponds to your second fresh plate. Deposit and spread the liquid as described in steps 2 and 3.

Identification

- A Escherichia coli
- B Serratia marcescens
- C Micrococcus luteus
- D Bacillus subtilus



5. Stack your group's of plates on top of one another and tape them together.

The plates should be left in the upright position to allow the cell suspension to be absorbed by the agar.

- 6. Place the set of plates in a safe place designated by your instructor.
- After the cell suspension is absorbed by the agar, you or your instructor will place the plates in the <u>inverted</u> position (agar side on top) in a 37° C bacterial incubation oven for overnight incubation (15-20 hours).

If the cells have not been absorbed into the medium, it is best to incubate the plates upright. The plates are inverted to prevent condensation on the lid, which could drip onto the culture and may interfere with experimental results.



Experiment



NOTE: Ensure the entire plate has been completely streaked over with the inoculating loop.





NOTE:

PREPARATION OF BACTERIAL LYSATES

Second Day

Micrococcus luteus may grow more slowly than the other microbes. If growth is not evident, continue incubation until a lawn of cells can be seen.

All the plates should have a confluent mass (lawn) of cells, not colonies. Bacterial lawns may have clumps and aggregates while others may have clearer zones surrounded by heavier growth. Extend the incubation period if sufficient growth is not evident.

- 1. Label two 10 ml test tubes with the letters corresponding to your plates and your group number. Label two screw cap microcentrifuge tubes with the letters corresponding to your plates and group number. Determine their empty weight to the nearest milligram (0.001g).
- 2. Add 4 ml of tris-EDTA-glucose (TEG) to each plate.
- 3. Using a sterile loop for each plate, scrape the cells from the agar surface by streaking back and forth.
- 4. Tilt the plate slightly. Using a 5 ml pipet with a suction device, pipet the cells up and down to break up larger clumps. Eject the liquid against the tilted agar surface to wash down residual cells. Use a different pipet for each plate.
- 5. Transfer all the resuspended cells (2.5 to 3.5 ml) from each of the plates to the corresponding 10 ml test tube. Mix the tube by vortexing or by other agitation to break up most of the remaining cell aggregates.
- 6. Mix the cells to obtain an even suspension. Using fresh pipets with a suction device, transfer 1 ml of the cells to the corresponding screw cap microcentrifuge tubes. Cap each of the tubes.
- 7. Place your microcentrifuge tubes in the microfuge, counter-balanced and centrifuge for 5 minute at full speed.
- 8. Remove all of the supernatant from the tubes with a pipet. Discard the supernants in a waste beaker.
- 9. Invert the tubes and blot residual liquid with a wipe or paper towel.
- 10. Determine the wet mass of the cell pellets.

NOTE: The wet mass of cell pellets may vary among the bacteria. It is recommended that the cell pellets be comparable in weight to ensure a consistent intensity of the protein bands in the final results. Suggested wet mass is between 0.020 and 0.025 grams.



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11. Resuspend the cell pellets with Tris-EDTA-Glucose (TEG) buffer (Component E) to approximately 100 mg cells per ml. The volume of buffer to add is given by the cell mass in grams multiplied by 10. For example, if the cell pellet had a mass of 0.025 gm (25 mg), then 0.25 ml of buffer should be added to the tube.

Resuspend the pellet by vigorous vortexing and agitation with a pipet. Mechanically dislodge the pellet under the buffer with a pipet if necessary.

- 12. Add one-tenth the volume of lysozyme solution to the cell resuspensions (e.g. 25 μ l to 0.25 ml). Mix. Incubate in a 37° C water bath for 30 minutes.
- 13. Add three times the volume of protein sample buffer (contains SDS and 2 mercaptoethanol) to the cell suspensions (e.g. 0.75 ml of buffer to 0.25 ml of cells, volume of lysozyme need not be considered).
- 14. Tighten the screw caps on the tubes. Suspend your sample tubes in a boiling water bath for 10 minutes.
- 15. Allow the tubes to cool. Centrifuge at full speed, 5 minutes in a microfuge.
- 16. Transfer most of the supernant to a fresh tube for storage. This is your protein lysate. Pellets may not be clearly visible in the tubes that were centrifuged. Set these tubes aside.



OPTIONAL STOPPING POINT

The crude protein samples may be kept refrigerated for a week. For longer storage, samples should be frozen but must be boiled again after thawing.





Optional Dilution of Protein Lysates

The yield of the protein lysates can vary among student groups and greatly depends on how well the preparation of the bacterial lysates is performed.

Prior to the protein denaturing procedure (below), teachers may wish to compare the final results by assigning various student groups to set up different dilutions of the protein lysates using the Protein Sample Buffer (Component G).

Suggested dilution range is 25% to 50% dilution of the protein supernatant collected in step 21 (page 10). For example, if a 25% dilution is required for a protein lysate, 3 parts of protein supernatant should be added to the one part of the Protein Sample Buffer.

Protein Denaturation

Denatured proteins tend to form super-molecular aggregates and insoluble particulates. Heating disrupts aggregates of denatured proteins.

- If the Unknown (component H) has already been rehydrated and heated by your lab instructor, proceed to denaturing the protein lysates as instructed in steps 1-2 below.
- If the Unknown (component H) has NOT been rehydrated and heated by your lab instructor, proceed to denaturing the Unknown, along with the protein lysates as instructed in steps 1-2 below.
- 1. Bring a beaker of water, covered with aluminum foil, to a boil. Remove from heat.
- 2. Suspend tubes containing the Unknown (Component H) along with the protein lysates (and/or their dilutions) in a boiling bath. Make sure the tubes are tightly capped and well labeled. The bottom of the tubes should be pushed through the foil and immersed in the boiling water for 5 minutes. The tubes should be kept suspended by the foil.
- 3. Proceed to loading the gel while the samples are still warm.

NOTE: Upon completion of loading the samples for electrophoresis, the unused portions of the protein samples can be frozen. Remove the samples from the freezer and follow steps 1-3 above to re-heat and run the samples when using them at a later time.





Electrophoresis of Proteins

PREPARING THE POLYACRYLAMIDE GEL FOR ELECTROPHORESIS

Precast Polyacrylamide Gels:

Precast polyacrylamide gels will vary slightly in design. Procedures for their use will be similar.

1. Open the pouch containing the gel cassette with scissors. Remove the cassette and place it on the bench top with the front facing up.

Note: The front plate is smaller (shorter) than the back plate.

- 2. Some cassettes will have tape at the bottom of the front plate. Remove all of the tape to expose the bottom of the gel to allow electrical contact.
- 3. Insert the Gel Cassette into the electrophoresis chamber.
- 4. Remove the comb by placing your thumbs on the ridges and pushing (pressing) upwards, carefully and slowly.

PROPER ORIENTATION OF THE GEL IN THE ELECTROPHORESIS UNIT

- 1. Place the gel cassette in the electrophoresis unit in the proper orientation. Protein samples will not separate in the gel if the cassette is not oriented correctly. Follow the directions accompanying the specific apparatus.
- 2. Add the diluted buffer into the chamber. The sample wells and the back plate of the gel cassette should be submerged under buffer.
- 3. Rinse each well by squirting electrophoresis buffer into the wells using a transfer pipet.

The gel is now ready for practice gel loading or sample loading.



The figure above shows a polyacrylamide gel cassette in the EDVOTEK® Vertical Electrophoresis Apparatus, Model #MV10.





Electrophoresis of Proteins



EDVOTEK® Cat. #638, Fine Tip Micropipet Tips are recommended for loading samples into polyacrylamide gels. A regular microtip may damage the cassette and result in the loss of protein samples.

PRACTICE GEL LOADING

EDVOTEK® Cat. #638, Fine Tip Micropipet Tips are recommended for loading samples into polyacrylamide gels. A regular microtip may damage the cassette and result in the loss of protein samples.

- 1. Place a fresh fine tip on the micropipet. Aspirate 20 μl of practice gel loading solution.
- 2. Place the lower portion of the fine pipet tip between the two glass plates, below the surface of the electrode buffer, directly over a sample well. The tip should be at an angle pointed towards the well. The tip should be partially against the back plate of the gel cassette but the tip opening should be over the sample well, as illustrated in the figure on page 13.

Do not try to jam the pipet tip in between the plates of the gel cassette.

4. Eject all the sample by steadily pressing down on the plunger of the automatic pipet.

Do not release the plunger before all the sample is ejected. Premature release of the plunger will cause buffer to mix with sample in the micropipet tip. Release the pipet plunger after the sample has been delivered and the pipet tip is out of the buffer.

5. Before loading protein samples for the actual experiment, the practice gel loading solution must be removed from the sample wells.

Do this by filling a transfer pipet with buffer and squirting a stream into the sample wells. This will displace the practice gel loading solution, which will be diluted into the buffer and will not interfere with the experiment.



Electrophoresis of Proteins

LOADING PROTEIN SAMPLES

Change fine pipet tips between loading each sample. Make sure the wells are cleared of all practice loading solution by gently squirting electrophoresis buffer into the wells with a transfer pipet.

Two groups will share each gel. The protein samples should be loaded in the following manner:

Group A

Lane 1	20 µl of unknown (H)
Lane 2	15 µl of <i>E. coli</i> (A)
Lane 3	15 µl of S.marcescens (B)
Lane 4	15 µl of <i>M.luteus</i> (C)
Lane 5	15 µl of <i>B.subtilus</i> (D)

Group B

Lane 6	20 µl of unknown (H)
Lane 7	15 μl of <i>Ε. coli</i> (A)
Lane 8	15 µl of S.marcescens (B)
Lane 9	15 μl of <i>M.luteus</i> (C)
Lane 10	15 µl of <i>B.subtilus</i> (D)

QUICK REFERENCE:

Protein Samples:

- A. E.coli
- B. S.marascens
- C. M.luteus
- D. B.subtilus
- H. Unknown

RUNNING THE GEL

- 1. After the samples are loaded, carefully snap the cover all the way down onto the electrode terminals. The black plug in the cover should be on the terminal with the black dot.
- 2. Insert the plug of the black wire into the black input of the power supply (negative input). Insert the plug of the red wire into the red input of the power supply (positive input).
- Set the power supply at the required voltage and run the electrophoresis for the length of time as determined by your instructor. When the current is flowing, you should see bubbles forming on the electrodes. The sudsing is due to the SDS in the buffer.
- 4. After the electrophoresis is finished, turn off power, unplug the unit, disconnect the leads and remove the cover.

Т	ime and Vo	Itage
Volts	Recomme Minimum	nded Time Optimal
125	60 min	75 min







Staining the Gel



STAINING WITH PROTEIN INSTASTAIN® IN ONE EASY STEP

EDVOTEK features a state-of-the-art, proprietary stain for DNA or Protein gels called InstaStain[®]. Protein Polyacrylamide gels can be stained with Protein InstaStain[®] cards in one easy step. Staining is rapid, sensitive and Polyacrylamide gels are ready for visualization in 1-3 hours.

InstaStain® Blue and InstaStain® Ethidium Bromide are also available from EDVOTEK for staining of DNA gels.

- 1. After electrophoresis, turn off the power and remove the gel cassette from the gel electrophoresis apparatus.
- 2. To remove the gel from the cassette, lay the cassette down and carefully remove the front plate by placing a coin or a spatula in the slot at the top edge, near the sample wells, and twist to separate the two plates of the cassette.
- 3. Gently lift the front plate away from the larger back plate. In most cases, the gel will stay on the back plate. If the gel partially sticks to the front plate, let it fall onto the back plate.
- 4. Pour approximately 100 ml of fixative solution in a small tray.
- 5. Transfer the back plate of the cassette (with the gel) into the tray containing the fixative solution. Wet gloved fingers with fixative solution and gently nudge the gel off the back plate and remove the plate, leaving the gel submerged in the fixative solution.
- 6. Gently float a sheet of Protein InstaStain® card with the stain side (blue) facing in the liquid. Remove the Protein InstaStain® card after 30 minutes.
- 7. Cover the staining tray with saran wrap to prevent evaporation.
- 8. Gently agitate on a rocking platform for 1-3 hours or overnight.
- 9. After staining, Protein bands will appear medium to dark blue against a light background* and will be ready for excellent photographic results.
- * Destaining is usually not required but can be carried out if the gel background is too dark. Gels can be destained in several changes of fresh destaining solution until the appearance and contrast of the protein bands against the background improves.



NOTE:

50 ml

10 ml

40 ml

Polyacrylamide gels are

very thin and fragile. Use care in handling to

avoid tearing the gel.

Fixative and Destaining

Solution for each gel (100ml)

Methanol

Glacial Acetic Acid

Distilled Water

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Staining the Gel

Storing the Gel

Once satisfactory result is achieved, the gel can be stored in distilled or deionized water.

For permanent storage, the gel can be dried between two sheets of cellophane (saran wrap) stretched in an embroidery hoop. Air-dry the gel for several days until the gel is paper thin. Cut the "extra" saran wrap surrounding the dried gel. Place the dried gel overnight between two heavy books to avoid curling. Tape it into a laboratory book.

Ordering Information:

InstaStain® Blue

Cat. #2003 for 40 gels, 7 x 7 cm Cat. #2004 for 100 gels, 7 x 7 cm Cat. #2006 Roll, 14 x 350 cm

InstaStain® Ethidium Bromide

Cat. #2001 for 40 gels, 7 x 7 cm Cat. #2002 for 100 gels, 7 x 7 cm Cat. #2005 Roll, 14 x 350 cm

InstaStain® Protein

Cat. #2016 for 15 gels, 10 x 10 cm Cat. #2017 for 30 gels, 10 x 10 cm





Study Questions

- 1. Can similar SDS polyacrylamide electrophoretic profiles be definitive criteria for establishing genetic relationships?
- 2. Are there technical problems you can think of that may cause an incorrect match or a missed match between knowns and unknowns using SDS polyacrylamide gel electrophoresis?
- 3. Do you know of any electrophoretic methods that could make identification of unknown protein lysates more accurate? (Information is not included in the introduction.)



Bacterial Protein Fingerprinting



Experiment

Instructor's Guide



Visit our web site for information about EDVOTEK's complete line of experiments for biotechnology and biology education.

HOW THIS EXPERIMENT IS ORGANIZED

This experiment module contains biologicals and reagents for six (6) groups sharing three (3) polyacrylamide gels (2 groups per gel). Enough buffer is included for three (3) vertical electrophoresis units (Model MV-10 or equivalent). Additional electrophoresis buffer is required for more than three units.

Note: Polyacrylamide gels are not included. You may choose to purchase precast gels (Cat. #s 651 or 652).

The experimental procedures consist of three major parts:

- 1) separation of proteins on polyacrylamide gels,
- 2) staining of protein bands,

3) identifying major protein bands in various extracts.

The staining of protein bands can be conducted using Protein InstaStain®, a new state-of-the-art method of staining. Protein InstaStain® is a proprietary staining method available exclusively from EDVOTEK®. You may wish to compare the staining methods by assigning various student groups to use different staining methods.



APPROXIMATE TIME REQUIREMENTS FOR PRE-LAB AND EXPERIMENTAL PROCEDURES

- 1. Pre-lab preparations will require approximately 20 minutes on the day of the lab.
- 2. Students will require approximately 15 minutes to heat samples and load the gel. Practice gel loading may require an additional 15 minutes if performed the same day of the lab.
- 3. Electrophoresis will require approximately 1 to 1.5 hours, depending on the voltage.

PRACTICE GEL LOADING

This experiment kit contains practice gel loading solution. If your students are unfamiliar with vertical gel electrophoresis, it is suggested that they practice loading the sample wells before performing the actual experiment.

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Bacterial Protein Fingerprinting

Experiment

Pre-Lab Preparations



PREPARATION OF AGAR PLATES Within one week of lab

Student participation is suggested if possible. Students can prepare agar plates on the first day of the lab. However, plates should be allowed to solidify for an hour before streaking.

- 1. Loosen, but do not remove, the cap on the ReadyPour[™] medium bottle. This will allow for the venting of steam during heating.
- Place the bottle in a boiling (100° C) water bath or oven to completely melt the ReadyPour medium. Using a hot glove, occasionally VERY GENTLY swirl to expedite melting.

Alternatively, you may microwave at <u>medium</u> setting to heat the bottle. Using a hot glove, occasionally VERY GENTLY swirl the bottle. **The ReadyPour medium will boil out of the bottle if left unattended.**

- 3. Allow the ReadyPour medium to cool to approximately 55° C (warm to the touch but not burning hot).
- 4. While the ReadyPour medium is cooling, label a total of 12 petri plates on their bottom halves:

3 Plates for A - Escherichia coli
3 Plates for B - Serratia marcescens
3 Plates for C - Micrococcus luteus
3 Plates for D - Bacillus subtilus

- 5. When the Ready Pour has cooled to approximately 55° C transfer approximately 12 ml of the medium to each of the 12 plates using a sterile 10 ml pipet. Rock the plate back and forth while adding the medium to obtain complete coverage.
- 6. Re-cover the plates and allow the medium to solidify.
- 7. Tape the plates and store them, inverted in the refrigerator. The plates should be used within 1 week.



ovens or hot plates.





Experiment

Pre-Lab Preparations

BACTERIAL SLANTS

First day, within 1 hour before the lab.

- 1. Using a different, sterile 1 ml pipet for each slant, add 1 ml of nutrient broth to slants A-D. Wait for 5 to 10 minutes before proceeding to step 2.
- 2. Use a sterile loop to enter slant A and thoroughly rub the entire immersed surface of the agar slant so the cells are released into the liquid. Close the vial and gently mix.
- 3. Using fresh loops, repeat step 2 for each of the slants B-D using a new sterile loop for each slant.

General comments for the first day of lab and group assignments.

Equilibrate a 37° C incubation oven. The slants will be passed between student groups.

Groups 1, 3 and 5:	(2) Plates and slant samples A, B
Groups 2, 4 and 6:	(2) Plates and slant samples C, D

If the students do not prepare the agar plates, day 1 of the lab will be relatively short (approximately 30 minutes to 1 hour). Plan accordingly.

RECONSTITUTION OF UNKNOWN SAMPLE (COMPONENT H) Second day

- 1. Add 130 µl of distilled or deionized water to the tube containing the Unknown (H) and allow the sample to hydrate for several minutes. Vortex or mix vigorously.
- Bring a beaker of water, covered with aluminum foil, to a boil. Remove from heat. Make sure the sample tube is tightly capped and well labeled. The bottom of the tube should be pushed through the foil and immersed in the boiling water for 5 minutes. The tube should be kept suspended by the foil.
- 3. Aliquot 20 µl of sample per student group. Have students load the sample onto the polyacrylamide gel with it is still warm to avoid aggregation.
- 4. Store any unused portion of reconstituted sample at -20° C and repeat steps 2 and 3 when using samples at a later time.

RECONSTITUTION OF UNKNOWN SAMPLE (COMPONENT H) Second day

Thaw the Protein Sample Buffer (G) at room temperature.

- 1. Using a permanent marker, label six (6) microcentrifuge tubes "buffer".
- 2. Aliquot 3 ml of Protein Sample Buffer (G) to each tube.
- 3. Distribute one "buffer" tube per student group.



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Instructor's Guide

Experiment

PreLab Preparations

LYSOZYME Second day, within 1 hour of lab.

- 1. Transfer 10 ml of cold TEG buffer (E) to a clean beaker, flask or test tube.
- 2. Add all the contents of tube F (lysozyme) to the buffer. Mix.
- 3. Store the enzyme solution on ice.
- 4. Aliquot 1.5 ml for each group.
- 5. Equilibrate a 37° C water bath. Have extra centrifuge tubes on hand for counter balances.

Notes Regarding Day Two Activities:

Sterility is not required for preparing the lysate. The volume of samples generated by the students will probably be much greater than the volumes needed for electrophoresis. These samples can be stored long term in the freezer. More gels can be purchased or prepared and several electrophoresis labs demonstrating protein lysates can be performed.

PREPARING ELECTROPHORESIS BUFFER

1. Prepare the electrophoresis buffer by adding and mixing 1 part Tris-Glycine-SDS 10x buffer concentrate to 9 parts distilled water.

The approximate volume of 1x electrophoresis buffer required for EDVOTEK Protein Vertical Electrophoresis units are listed in the table below. The buffer should just cover the back plate of the gel cassette.

Tris-Glycine-	SDS Electropho	oresis (Chambe	er) Buffer
EDVOTEK Model #	Concentrated Buffer (10x)	Distilled + Water :	Total = Volume
MV-10	58 ml	522 ml	580 ml
MV-20	95 ml	855 ml	950 ml



EDVOTEI



Pre-Lab Preparations

ELECTROPHORESIS TIME AND VOLTAGE

Your time requirements will dictate the voltage and the length of time for your samples to separate by electrophoresis. Approximate recommended times are listed in the table at right.

PREPARING STAINING AND DESTAINING SOLUTIONS

The stock solution is used for staining and destaining with Protein InstaStain®

- 1. Solution for staining with Protein InstaStain®
 - Prepare a stock solution of Methanol and Glacial Acetic Acid by combining 180 ml Methanol, 140ml Distilled water, and 40ml Glacial Acetic Acid.

2. Destaining Solution

• Use the stock solution of Methanol, Glacial Acetic acid and distilled water (in Step 1) to destain the gel(s).

Т	ime and Vo	Itage
Volts	Recomme Minimum	nded Time Optimal
125	60 min	75 min

EDVOTEK

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Instructor's Guide

Experiment

Idealized Schematic of Results

All protein lysates will contain many closely spaced bands. The molecular weight range of the lysate proteins extend from over 95,000 to below 14,000. Unknown (Component H) is a protein lysate from *E.coli*.

The gel below is an example of the different dilutions of the extracted protein lysates. A 25% dilution of the protein lysates is shown in lanes 2-5, and a 50% dilution of the protein lysates is shown in lanes 7-10.

Lanes	<u>Sample</u>
-------	---------------

- 1 Unknown (Component H)
- 2 E. coli
- 3 S.marcescens
- 4 M.luteus
- 5 B. subtilus

Lanes Sample

- 6 Unknown (Component H)
- 7 E. coli
- 8 S.marcescens
- 9 M.luteus
- 10 B. subtilus





Please refer to the kit insert for the Answers to Study Questions



Material Safety Data Sheets Full-size (8.5 x 11") pdf copy of MSDS is available at www. edvotek.com or by request.

EDVOTEK	Material Safety Data Sheet May be used to compay with OSHA's Hazard Communication Sandard. 29 CFH 1910.1200 Standard must be consulted for specific requirements.	Material Safety Data Sheet Maybe used to comply with OSIAX: Hazard Communication Standard. 29 CR 1910 (200 Standard must be consulted for Specific requirements.	Material Safety Data Sheet Nav be used to comply with GSHA's Hazad Communication Standard. 29 CFR 1910.1200 Standard must be consulted for specific requirements.
IDENTITY (As Used on Label and List) Protein InstaStain	Note. Blank spaces are not permitted. If any item is not applicable, or no information is available, the space must be marked to indicate that.	DEMITY (As Used on Label and List) Here: Bink rows are soperated. If any term in not applicable are conformation is validable, the space must be marked to indicate that.	IDENTITY (As Used on Label and List) Now: Blank spaces are not permitted. If any item is not Tris-Glycine SDS Running Buffer (IOX) Baptedia on the material on a standard, the space must
Section I Manufacturer's Name EDVOTEK, Inc. Address (Number, Street, City, State, 2 1121 5th Street NW	Energency Telephone Number (202) 370-1500 Telephone Number for Information Date Prepared (1,0,1,1)	Section 1 Manufacturer's Name Emergency Telephone Number EDVOTEK, Inc. Address Number Street, City, State, Zip Code) Address Number Street, City, State, Zip Code) Date Prepared 1121 5th Street NW	Section I Emergency Telephone Number (202) 370-1500 Manufacturer's Name Emergency Telephone Number (202) 370-1500 Address (Number Carbon (202) 370-1500 Telephone Number (202) 370-1500 Address (Number Carbon (202) 370-1500 Date Prepared
Washington DC 20001	5gnature of Preparer (optional)	Washington DC 20001 Signature of Preparer (optional)	Washington DC 20001 Signature of Preparer (optional)
Section II - Hazardous Ingredi Hazardous Components [Specific Chemical Identity: Common Name(s)] Methanol (Methyl Alcohol) 200ppr CH3OH	ents/dentify Information Cher Limits CSHA PEL ACGH TLV Recommended % (Optional) 1 200ppm No data 90%-100%	Section II - Hazardous Ingredients/Identify Information Hazadous Components Specific OSHAPEL ACGHTU Recommended % (Optional Chemistalidentity: ComonNamec) This product contains no hazardous materials as defined by the OSHA Hazard Communication Standard.	Section II - Hazardous Ingredientis/Identify Information Hazardous Components (Specific Chemical Identify: Common Names) OSHA P.EL ACGIH TLV Refer Limits Ths CXAP P.EL ACGIH TLV Refer Limits 95 Observe Use Strate State
Section III - Physical/Chemica	l Characteristics	Section III - Physical/Chemical Characteristics	Section III - Physical/Chemical Characteristics
Boiling Point	65°C Specific Gravity (H20 = 1) .79	Boiling Point No data Specific Gravity (H ₂ 0 = 1) No data	Boiling Point No data Specific Gravity (H20 = 1) No data
Vapor Pressure (mm Hg.) Vapor Density (AIR = 1)	96mmHg Metting Point NVA Evaporation Rate A 6	Vapor Presure (mm Hg.) No data Melting Point No data Vapor Pressure (mm Hg.) No data (Bayon Accessere = 1) No data	Vapor Pressure (mm Hg) No data Melting Point No data Vapor Poresture (mm Hg) No. data Evaporation Rate No data
Solubility in Water Complete (10)	(Laury) Avenaus = 1)	Solubility in Water Soluble	Solubility in Water Soluble
Appearance and Odor chemical bour	nd to paper, no odor	Appearance and Odor Blue liquid, no odor	Appearance and Odor Clear, no odor
Section IV - Physical/Chemics Flash Point (Method Used) (closed cup) 12	I Characteristics LEL UEL P*C Flammable Limits LEL UEL	Section IV - Physical/Chemical Characteristics Flash Point (Method Used) No data Flammable Limits LEL No data No data	Section IV - Physica/Chemical Characteristics Pash Point (Method Used) No data No data No data
Extinguishing Media Use alcohol fo:	am, dry chemical or carbon dioxide. (Water may be ineffective)	Extinguishing Media Dry chemical, carbon dioxide, water spray or foam	Extinguishing Media Water spray, carbon dioxide, dry chemical powder or appropriate foam
Special Fire Fighting Procedures Wear SCBA with full facepiece Move contrinent from fireares	operated in positive pressure mode.	Special Fire Fighting Procedures Use agents uitable for type of surrounding fire. Keep upwind, avoid breathing hazardous sulfur oxides and bromides. Wear SCBA.	Special Fire Fighting Procedures Wear SCBA and protective clothing
Unusual Fire and Explosion Hazards Close containers exposed to heat n	Vapors may flow along surfaces to distant ignition sources. nay explode. Contact w/ strong oxidizers may cause fire.	Unusual Fire and Explosion Hazards Unknown	Unusual Fire and Explosion Hazards May emit toxic fumes
Section V - Reactivity Data		Continue V - Desertivities Data	Section V - Reactivity Data
Stability Unstable Stable	Conditions to Avoid X None	Section v - reactivity batia Stability - Instable Conditions to Avoid Stability - Nore	Stability Unstable Conditions to Avoid Stable X Strong oxidizing agents, strong acids
Incompatibility	Strong oxidizing agents	Incompatibility None	Incompatibility Strong oxidizing agents
Hazardous Decomposition or Byproducts	Carbon monoxide, Carbon dioxide, Sulfur oxides	Hazardous Decomposition or Byproducts Sulfur oxides, and bromides	Hazardous Decomposition or Byproducts Carbon monoxide, carbon dioxide, sulfur oxides, sodium oxides
Hazardous May Occur Polymerization Mill Not Occ	Canditions to Avoid Sur X None	Hazardous May Occur Conditions to Avoid Polymerization Will Not Occur X None	Hazardous May Occur Oorditions to Avoid Polymerization Will Home May Occur X Scenition VI - Lanoitth Learnal And Deals
Boute(s) of Entry: Index Index	tation? Skin? Ingestion?	Section VI - Health Hazard Data Route(s) of Entry: Inhalation? Vac Skin? Vac Ingestion? Vac	Route(s) of Entry: Ingestign? Skin? Skin? Ingestign?
Health Hazards (Acute and Chronic) Ir Chronic exposure may cause lung damag	ritating to eyes, skin, mucous membranes and upper respiratory tract, ge or pulmonary sensitization	Health Hazards (Acute and Chronic) Acute eye contact: May cause irritation.	Health Hazards (Acute and Chronic) May cause irritation to eyes, skin, and mucous membranes.
Carcinogenicity: No da No da	P? IARC Monographs? OSHA Regulation? tta No data No data	Carcinogenicity: NTP? IARC Monographs? OSHA Regulation? No data available	Carcinogenicity: NTP? IARC Monographs? OSHA Regulation? No data No data No data No data
Signs and Symptoms of Exposure Respiratory tract: burning sensation. C	oughing, wheezing, laryngitis, shortness of breath, headache	Signs and Symptoms of Exposure May cause skin or eye irritation	Signs and Symptoms of Exposure Irritation
Medical Conditions Generally Aggravate No data	d by Exposure	Medical Conditions Generally Aggravated by Exposure None reported	Medical Conditions Generally Aggravated by Exposure Unknown
Emergency First Aid Procedures Flush skin/eyes w/ large amounts of w of water or milk. Do not induce vomiti	ater. If inhaled, remove to fresh air. Ingestion: give large amounts ng.	Emergency First Aid Procedures Treat symptomatically and supportively. Rinse contacted area with copious amounts of water.	Ernergency First Aid Procedures Skin/cye contact: flush w/ water. Inhalation: remove to fresh air. Ingestion: Seek medical attention
Section VII - Precautions for S	afe Handling and Use	Section VII - Precautions for Safe Handling and Use	Section VII - Precautions for Safe Handling and Use
Steps to be Taken in case Material is He. Evacuate area. Wear SCBA, rubber bo chemical incinerato equipped w/ an aft.	eased for Spilled ots and rubber gloves. Mop up w/ absorptive material and burn in erburner and scrubber.	Steps to be Taken in case Material is Released for Spilled Wear eye and skin protection and mop spill area. Rinse with water.	Steps to be Taken in case Material is Released for Spilled Wear protective clothing. Avoid contact. Mop up with absorbant material and dispose of properly.
Waste Disposal Method Observe all federal, state, and local law	2	Waste Disposal Method Observe all federal, state, and local regulations.	Waste Disposal Method Follow all state, federal, and local regulations.
Precautions to be Taken in Handling and Wear protective gear. Avoid contactin	Storing halation.	Precautions to be Taken in Handling and Storing Avoid eye and skin contact.	Precautions to be Taken in Handling and Storing Avvid contract known www.frum.heat
Other Precautions Strong sensitizer		Other Precautions None	Other Prezentions
Section VIII - Control Measure	ø	Section VIII - Control Measures	Section VIII - Control Measures
Respiratory Protection (Specify Type)	NIOSH/MSHA approved respirator	Respiratory Protection (Specify Type)	Respiratory Protection (Specify Type)
Ventilation Local Exhat	ust No Speciather furne hood (General) No Other None	Ventilation Local Exhaust Yes Special None Mechanical (General) Yes Other None	Ventilation Local Exhaust No Special None Mechanical (General) Yes Other None
Protective Gloves Rubber	Eye Protection Splash-proof goggles	Protective Gloves Yes Eye Protection Splash proof goggles	Protective Gloves Chem resistant Eye Protection Safety goggles
Other Protective Clothing or Equipment	Rubber boots	Other Protective Clothing or Equipment None required	Other Protective Clothing or Equipment Lab coat, coveralls
Work/Hygienic Practices	Avoid prolonged or repeated exposure	Work/Hygienic Practices Avoid eye and skin contact	Work/Hygienic Practices Prevent contact



elephone Number for information (202) 370-1500 No data (202) 370-1500 No data No data S RE tive individua armitted. If any item is available, the space istible solvent and burn in a chemical incinerator equipped w/ an afterburne Ingestion? Material Safety Data Sheet May be used to compay with OSHANA Hazard Communication Standard. 29 CFR 1910.1200 Standard must be consulted for specific requirements. ction Chem safety goggle: Special Fire Fighting Proceedures Wear SCBA and protective diothing to prevent contact w/ skin and eyes. data Ň 10-24-11 Telephone Number eye/skin contact: flush w/ large amounts of water Specific Gravity (H₂0 = 1) ACGIH TLV Material is Released for Spilled safety goggles, rubber boots and heavy rubber Note: Blank applicable, or be marked to Evaporation Ra (Butyl Acetate = Extinguishing Media Water spray, carbon dioxide, dry chemical powder Melting Point ions to Avo No data Skin? tion VIII - Precautions for Safe Handling and Use No data nergency lammable ay cause ABC Mor Eye F tion IV - Physical/Chemical Characteristics Clear liquid, no odor Avoid contact No data No data fedical Conditions Generally Aggravated by Exposure No data Lab coat ical Charact tion II - Hazardous Ingredients/Ide EDVOTEK, Inc. Address (Number, Street, City, State, Zp Code) Chem. resistant gloves and scrubber. Observe federal, state, and loc: scautions to be Taken in Handling and Storin Wear protective clothing, avoid contact Allergic Soluble 1121 5th Street NW Washington DC 20001 No data ction VI - Health Hazard Data nhala Section VIII - Control Measures Respiratory Protection (Specify Type) May Occur Other Protective Clothing or Equipment ENTITY (As Used on Label and List) Local Exha nusual Fire and Explosion Hazards ction 🔳 - Physical/Chem vater. ED VOTEK. (apor Pressure (mm Hg.) Taken in case A rator, chemical : /apor Density (AIR = 1) mouth /ork/Hygienic Practices None ance and Odor ards (Acute r's Nam olubility in Water Rinse ctive Gloves of Entry: dous erization toiling Point ction